## **Binary Ionic Compounds Questions**

- 1. a) In an ionic compound, what do you know about the sum of the charges on the positive ions compared to the sum of the charges on the negative ions?
- b) Calculate the sum of the ionic charges in the compound BaCl<sub>2</sub>. Show your calculation.
- 2. Draw a Bohr diagram to show the electron transfer that occurs when magnesium and fluorine form magnesium fluoride.
- 3. Write the names for the following compounds:
- a) KCl
- b) Na<sub>3</sub>P
- c) CaF<sub>2</sub>
- 4. In mining, some minerals are referred to as ferrous. What metallic element is present in these compounds? (Hint: Look at the letters that begin the word.)
- (P, D) 5. Calculate the sum of the ionic charges in the compound Al<sub>2</sub>O<sub>3</sub>. Show your calculation.